Data, data everywhere, nor any bit processable Opportunities for amalgamating and opening u hemical data and information relevant to hazar recognition and safety planning

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ease note ...

am an informatician
work with data .lots. of it
help to create public interfaces to chemical (biology) information
work with health and safety professionals, authoritative sources of H&S
nformation, (increasingly) those involved in H&S communities (including themistry librarians

am not a chemical health and safety (H&S) professional



pecial thanks

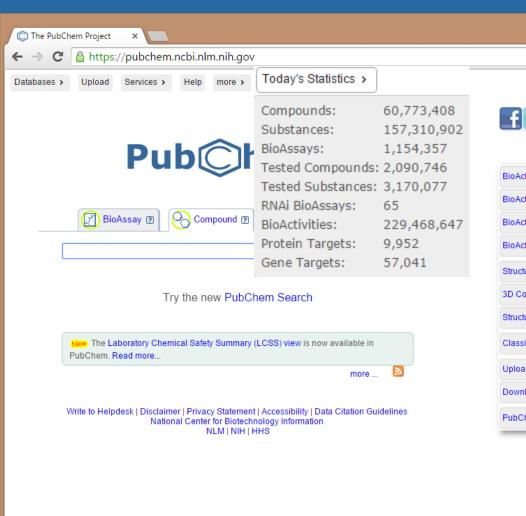
- Chemical Health and Safety collaborators
- Especially: Leah McEwen, Ye Li, Ralph Stuart
- Software collaborators
- Especially: Daniel Lowe and Roger Sayle (NextMove Software .. LeadMine)
- All Contributors and Collaborators
- This research was supported [in part] by the Intramural Research Program c he NIH, National Library of Medicine.

ubChem resource

Primary aim is to provide nformation known about chemical substances

- Contains authoritative resources
- Contains manually curated resources
- Includes health and safety
- There are errors
 - Authoritative doesn't mean error free
 - Curated doesn't mean error free
- We DO NOT curate (known errors reported, filtered)
- We can leverage curated content

https://pubchem.ncbi.nlm.nih.g





nemical Health and Safety Data Sources

ological Safety Data Sheets

NIOSH Pocket Guide to Chemical Hazards



http:// Does each organization (or scientist) use their own favorite data source(s)?



sment



national Chemical Sa

ıvironmental Hea

Do these various data sources provide consistent information (gaps, errors)?



SIRI MSDS Index



SDS and Chemi





Uni How does the health and safety decision change with different information (or lack of it)?

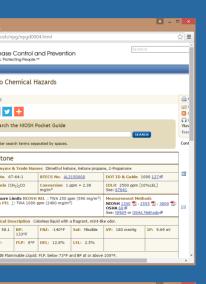


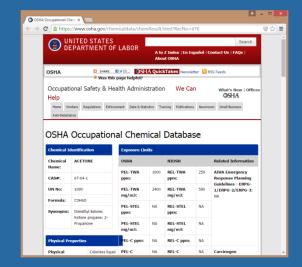
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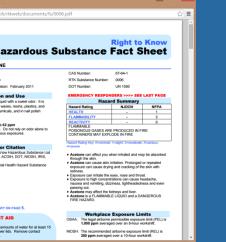
OXNET DATABASE



nemical Health and Safety Information







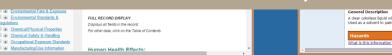
Acetone .. seven sources CDC NIOSH, NJ HSFS, DOL OSHA, WHO ICSC, NLM HSDB, EPA ATSDR, EPA/ NOAA CAMEO



One chemical .. many primary sou Each resource has some overlaps Each resource has unique content



What if you have ten chemicals?
What if you have ten sources for each?
How much time will one spend reading t



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ubChem Laboratory Chemical Safety Sheet (LCSS)

Pull in primary chemical health and safety resources (as-is possible)

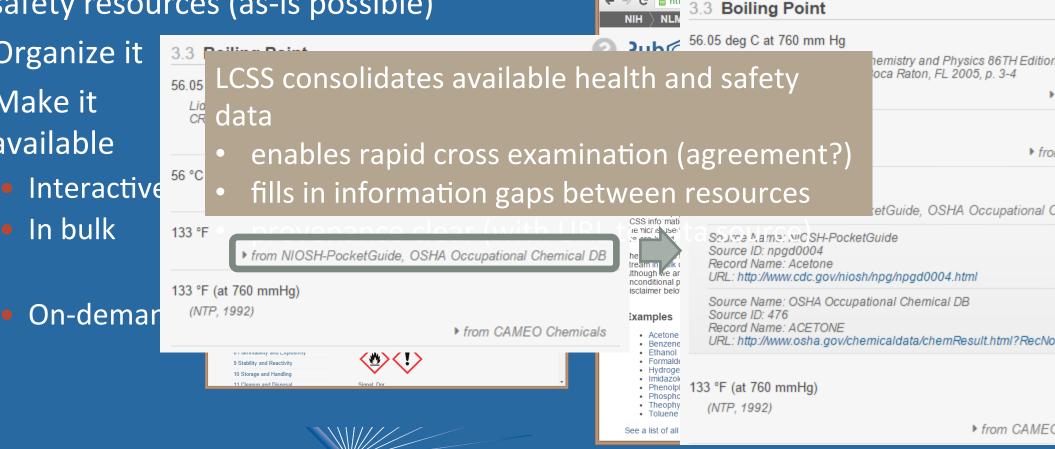
Make it

vailable

In bulk

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https://pubchem.ncbi.nlm.nih.gov/



ealth and Safety Information summaries are for human

reat! Succinct write-ups. What nore can one ask for?!

Part of a CAMEO record for Acetone

Fire Hazard

Excerpt from GUIDE 127 [Flammable Liquids (Polar / Water-Miscible)]:

HIGHLY FLAMMABLE: Will be easily ignited by heat, sparks or flames. Vapors may forn mixtures with air. Vapors may travel to source of ignition and flash back. Most vapors than air. They will spread along ground and collect in low or confined areas (sewers, be tanks). Vapor explosion hazard indoors, outdoors or in sewers. Those substances desig (P) may polymerize explosively when heated or involved in a fire. Runoff to sewer may explosion hazard. Containers may explode when heated. Many liquids are lighter than 2012)

Health Hazard

INHALATION: vapor irritating to eyes and mucous membranes; acts as an anesthetic concentrations. INGESTION: low order of toxicity but very irritating to mucous membranes prolonged excessive contact causes defatting of the skin, possibly leading to dermatitis 1999)

Reactivity Profile

It was reported that a mixture of ACETONE and chloroform, in a residue bottle, exploraddition of acetone to chloroform in the presence of base will result in a highly exothe it is thought that a base was in the bottle [MCA Case History 1661. 1970]. Also, Nitro sealed in a tube with a residue of acetone in the presence of platinum catalyst, gave a reaction [Chem. Eng. News 35(43):60. 1967]. The reaction of nitrosyl perchlorate an ignites and explodes. Explosions occur with mixtures of nitrosyl perchlorate and prima [Ann. Chem. 42:2031. 1909]. Reacts violently with nitric acid. Also causes exothermi when in contact with aldehydes.

uman understanding

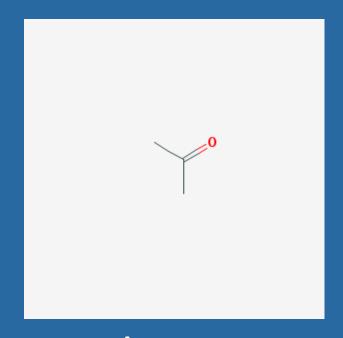
As a chemist, you can understand and recognize that this picture is the chemical acetone

ou can put a name or registry number next to it

s this not good enough?

The picture has no association with the name or registry number

The computer 'sees' a binary image



Acetone 67-64-1



omputer understanding

Give a computer a chemical

understand it

Computer cal he structure

o the structi

tructure a (r Computer understanding can help provide human understanding

If the computer understands, we Computer ca can leverage it for search, analysis, and more

Computer can generate other key nformation from structure

propan-2-one 58.07914 g/m

67



ealth and Safety Information summaries for computers

Can human readab nade computer un Potential workflow involving Computers and Humans

- Humans need to provide the terms and organize them
- Computer can recognize these terms (NER)
- Identify relationships between terms (Human/NLP)
- Relationships fed back to computer
- Harmony ensues .. better navigation, search, analysis

LATION: vapor irritating to eyes and mucous membranes; acts as an anesthetic in ver

th Hazard

entrations. INGESTION: low order of toxicity but very irritating to mucous membranes. nged excess Sounds good but how?

Computers are not very good at this (yet)

BUT computers combined with humans might be able to do something useful

rm, in a residue bottle, explod e will result in a highly exothe story 1661, 1970], Also, Nitro ce of platinum catalyst, gave a tion of nitrosyl perchlorate an nitrosyl perchlorate and prima ic acid. Also causes exothermi

lermatitis. (US

nen in contact with aidenvoes.

orking with chemical information has CAVEATS!

Chemical information is a bit of a mess and can be rather nuanced

- Names, names, and more names (210M in PubChem)
 - Some standard names are not open and cannot be used/verified without \$\$\$
 - Name/structure associations vary by use case (many overlapping)
 - Acetic acid vs. Acetic acid tri-hydrate
 - Formaldehyde: (gas) vs. Formalin (liquid , 40% formaldehyde w/ water)
 - Sulfuric acid: SO₃ (gas) vs. H₂SO₄ (liquid)
 - Glucose: L/D, ring open/closed, alpha/beta/both vs. Glucose monohydrate
 - Large corpus in the 'wild' .. data source dependent nuances

/erify with primary source(s) prior to information use

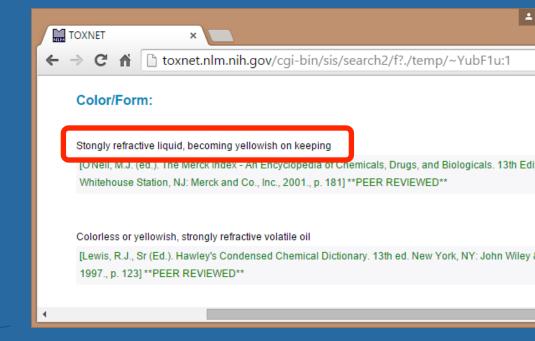
i.e., is this the form of the chemical I care about?



orking with chemical information has CAVEATS!

Chemical annotation in PubChem is text locked (i.e., meant for humans)

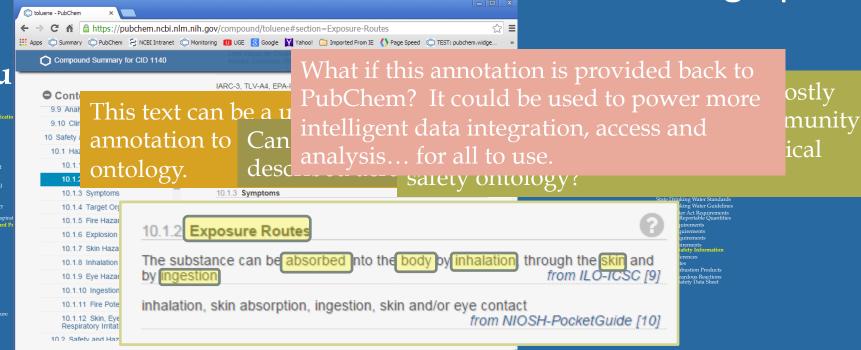
- Benzaldehyde
 - Color
 - Stongly refractive liquid, becoming yellowish on keeping
 - Colorless or yellowish, strongly refractive volatile oil
 - Boiling point
 - 179 deg C
 - 179 °C
 - 354 °F (at 760 mmHg)
 - Flash point
 - 145 deg F, 63 deg C (Closed cup)
 - 73.9 deg C (Open cup)
 - 63 °C



ow can PubChem help?

Well .. we have lots of data

Eureka!!!
Let's Make a connection graph of knowledge



/hat is a knowledge graph?

Knowledge graphs could be envisaged as a network of all kind things which are relevant to a specific domain or to an organization.

https://blog.semantic-web.at/2014/07/15/from-taxonomies-over-ontologies-to-knowledge-graphs/

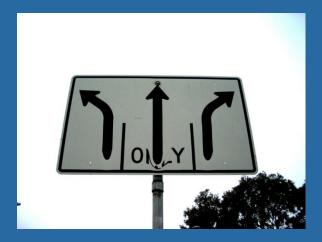




/hy a knowledge graph?

(To be) able to make complex queries over the entirety of all kind of information. By breaking up the data silos there is a high probability that query results become more valid.

https://blog.semantic-web.at/2014/07/15/from-taxonomies-over-ontologies-to-knowledge-graphs/





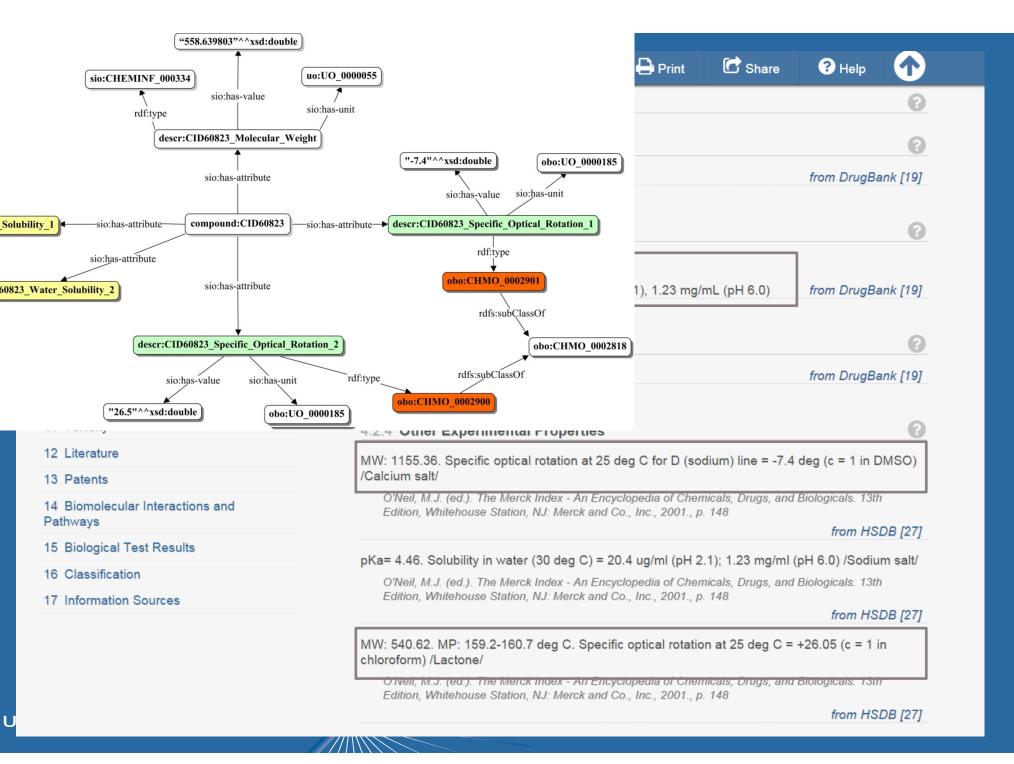
rend towards community based annotation

'Chemical Safety Ontology" (CSO) participants request and examine a provided table of contents of chemical safety information in PubChem to dentify areas of interest

PubChem provides text for a given area of interest to CSO to train the ontology by adding appropriate concepts:

- 1. PubChem uses current set of ontology terms to auto-annotate the text
- 2. CSO manually reviews the auto-annotation for ontological gaps, makes fixes
- 3. Iterate
- PubChem may be able to provide tools that help to automate this iterative process and could even serve as a locus for volunteers to semi-manually innotate chemical safety text
- PubChem may be able to act as a repository for annotation
- Integration into the PubChemRDF project





nemical Reactivity

Many scenarios of acute interest

- Reaction planning
 - Anything to worry about? Proper protective clothing? Explosion hazards?
 Incompatibilities? Risks? Storage? Solvent incompatibility? Etc.

Chemical reactivity classification

- NOAA Chemical Reactivity Worksheet (CRW)
- Brethericks
- Knowledge base variants CAMEO (CRW)

Nouldn't it be great to classify all of PubChem?

 Imagine tools where you could determine incompatibilities for any arbitrary chemical reaction and get an automated assessment



ummary

- PubChem provides chemical health and safety information LCSS for chemical
- Clear provenance of information provided
- Assembling health and safety information for easy access
- Breadth and depth of available information varies greatly per chemical
- Chemical information is (sadly) messy
- PLEASE double check the information that it deals with the same thing
- Norking with chemical health and safety community towards:
- Adding structure to textual data to make it computable
- Chemical reactivity classification



PubChem Crew ...

Steve Bryant

Jie Chen

Tiejun Chen

Gang Fu

Renata Geer

Asta Gindulyte

Lianyi Han

Jane He

Sigian He

Sunghwan Kim

Ben Shoemaker

Paul Thiessen

Jiyao Wang

Yanli Wang

Bo Yu

Jian Zhang

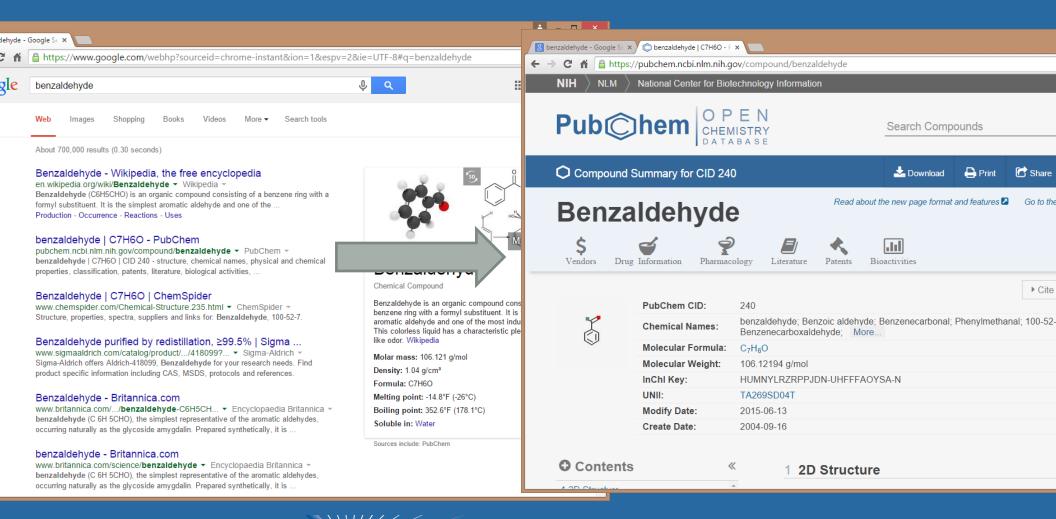
Special thanks to the NCBI Help Desk, especially Rana M



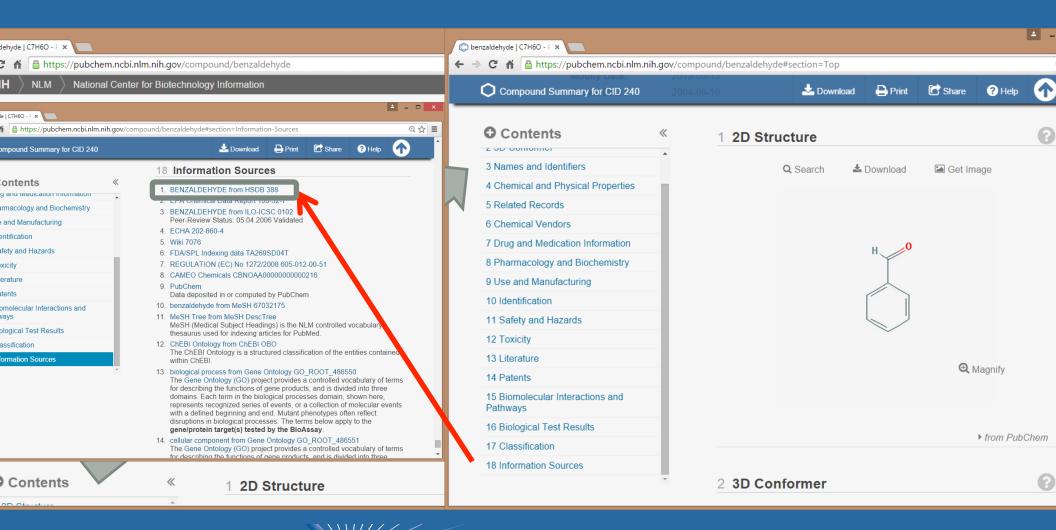
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nemical information is everywhere



enzaldehyde as a case study



PubChem Compound TOC - Experimental Properties

ysical Description

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ish Point

lubility

nsity

por Density

por Pressure

gP

8⁰

enrys Law Constant

mospheric OH Rate

nstant

bility

Optical Rotation

Auto-Ignition

Decomposition

Viscosity

Corrosivity

Volatility

Heat of Combustion

Heat of Vaporization

pН

Surface Tension

Evaporation Rate

Ionicity

Dispersion

Polymerization

Odor Threshold

Enthalpy of Formation

Entropy of Formation

Thermal Conductivity

Electrical Conductivity

Heat of Capacity

Bioavailability

Ghose Filter

Hydrophobicity

Isoelectric Point

Polarizability

Refractive Index

Caco2 Permeability

pKa

pKb

Dissociation Constants

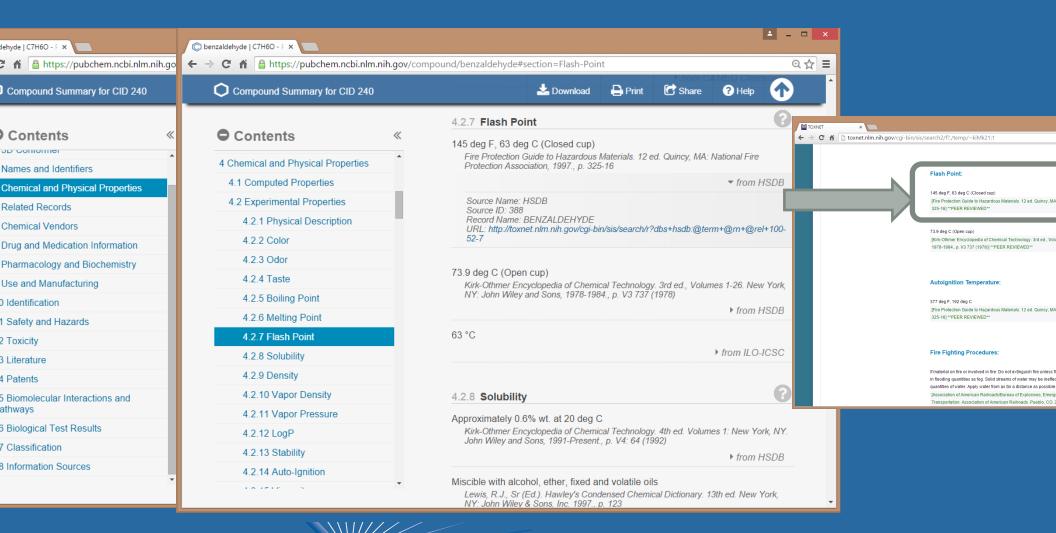
Relative Evaporation Rate

Other Experimental Properties

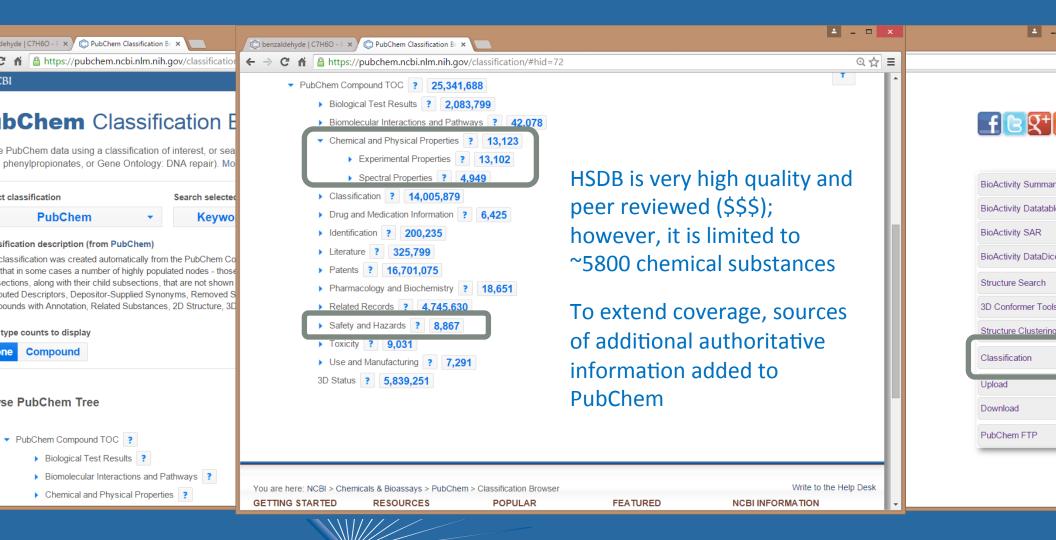
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enzaldehyde chemical and physical properties

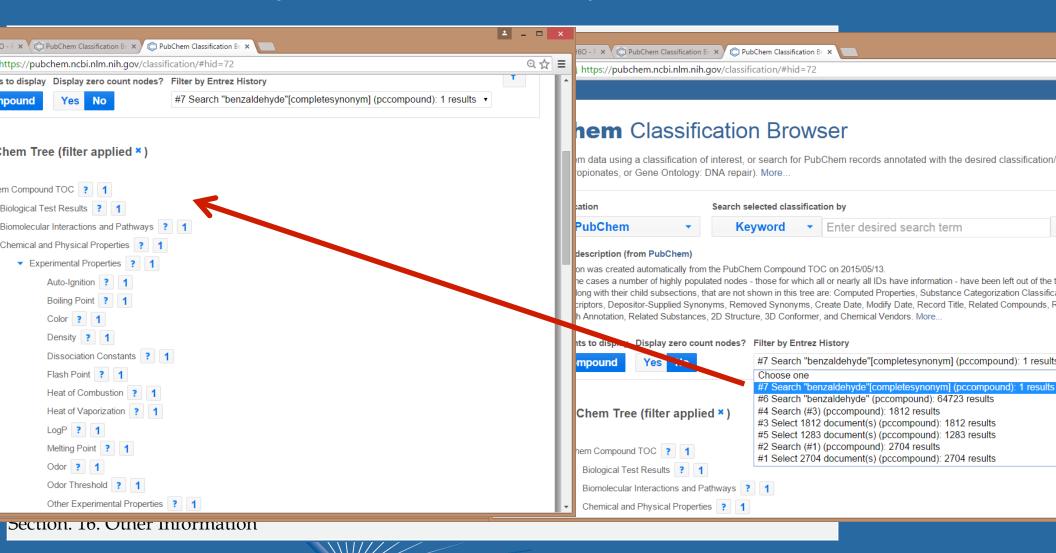


ubChem Compound Table of Contents



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OSHA SDS (Safety Data Sheet, formerly MSDS) standard



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